Synthesis of 4,12-Dihydro-4-oxoquino-[1,8a,8-a,b]quinoxaline-5-carboxylic Acid Derivatives

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The synthesis and antibacterial activity of 1-substituted amino-2-fluoro-4,12-dihydro-4-oxoquino[1,8a,8-a,b]quinoxaline-5-carboxylic acid derivatives are described. The synthetic route includes a carbon homologation and two intramolecular nucleophilic displacement cyclization reactions leading to the novel 4,12-dihydro-4-oxoquino[1,8a,8-a,b]quinoxaline-5-carboxylic acid heterocycle.

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In recent years, the 4-pyridone-3-carboxylic acid class of orally active antibacterials known collectively as the quinolones has stimulated a great deal of interest in the synthesis of 1,4-dihydro-4-oxoquinoline-3-carboxylic acid derivatives. This class of antimicrobial agents includes Norfloxacin 1 [1], Ciprofloxacin 2, Difloxacin 3 [3] and Ofloxacin 4 [4]. Compound 4 has an oxazine moiety. During our research on the synthesis of novel quinolone derivatives as antimicrobial agents, we synthesized a series of hybrid quinolone antibacterials containing the oxazine moiety. These compounds, 1-substituted amino-2fluoro-4-oxo-4H-quino[8,8a-1-b,c][1,4]benzoxazine-5-carboxylic acid derivatives 5, were found to possess antibacterial activity [5]. Since amino substitution in quinolone derivatives may improve the absorption or tissue distribution [6], compound 6, which has an NH group in place of the ring oxygen in 5, was chosen as a synthetic target.

The 4,12-dihydro-4-oxoquino[1,8a,8-a,b]quinoxaline-5-carboxylic acid (6) represents a novel heterocyclic ring system. In this paper, we would like to report the syntheses and antimicrobial evaluations of 1-(3-aminopyrrolidin-1-yl)-2,9-difluoro-4,12-dihydro-4-oxoquino[1,8a,8-a,b]quinoxaline-5-carboxylic acid (15) and 1-(4-methylpiperazin-1-yl)-2,9-difluoro-4,12-dihydro-4-oxoquino[1,8a,8-a,b]quinoxaline-5-carboxylic acid (16). Our synthetic pathway to the desired ring skeleton required two intramolecular

nucleophilic displacement cyclization reactions involving nitrogen as a nucleophile, a process that has been successfully applied to the syntheses of 1-aryquinolone [3], Amifloxacin [7] and benzoxazine derivatives [5]. Hence, compounds 15 and 16 were synthesized by a route illustrated in Scheme I.

Scheme I

$$COCH_{3} \longrightarrow F \longrightarrow CO_{2}H \longrightarrow F \longrightarrow COCH_{1}CO_{2}E$$

$$T \longrightarrow R_{1} \longrightarrow R_{2} \longrightarrow R$$

Oxidation and nitration of 2,4-dichloro-5-fluoroacetophenone (7) [8] was achieved in one reaction flask by reacting it with a mixture of concentrated sulfuric acid and nitric acid to yield the 2,4-dichloro-3-nitro-5-fluorobenzoic acid (8) (mp 188-191°, 82%). Treatment of this acid 8 with

thionyl chloride gave the corresponding acid chloride which, without purification, was reacted with the dilithiodianion of monoethyl malonate [9] to give ethyl 2,4-dichloro-3-nitro-5-fluorobenzoylacetate (9) as an oil (97%). It existed in both keto and enol forms. Treatment of this ester 9 with triethylorthoformate in acetic anhydride gave the one carbon homolog enol ether intermediate which was allowed to react with a slight excess of 2,4-difluoroaniline in methylene chloride at room temperature in give ethyl 3-(2,4-difluoroanilino)-2-(2,4-dichloro-3-nitro-5-fluorobenzoyl)acrylate (10). This enaminoketo ester exists in both trans and cis forms in solution with one of the isomers being predominant.

Regiospecific cyclization of the crude compound 10 with one molar equivalent of sodium hydride in tetrahydrofuran (THF) yielded ethyl 1-(2,4-difluorophenyl)-6fluoro-7-chloro-8-nitro-1,4-dihydro-4-oxoguinoline-3-carboxylate (11) (mp 189.5°, 79% overall from 9). A desired synthetic pathway to a variety of amino derivatives would require a common intermediate having the basic ring skeleton with a leaving group at the 1-position such as compound 17. Displacement of compound 17 at the 1-position with different amines would generate many amino derivatives. The compound 17 would, in theory, be derived from 11 by a reduction and cyclization synthetic sequence. The route described below, however, was chosen to take advantage of the activation of the chloro group by the α -nitro substituent, and thereby reduce the likelihood of displacement of the fluorine substituent. If the cyclization reaction were performed first to yield an intermediate such as 17, the NH group would have had a

deactivating effect on the chloro group; consequently, the displacement reaction was performed before cyclization.

Displacement of 11 with an excess of 3-acetamidopyrrolidine or 4-methylpiperazine in THF yielded the ethyl 1-(2,4-difluorophenyl)-6-fluoro-7-substituted amino-8nitro-1,4-dihydro-4-oxoquinoline-3-carboxylates 12a (mp 117°, 83%) or 12b (mp 204.5°, 73%). Reduction of 12a or 12b with Ranev nickel vielded the ethyl 1-(2.4-difluorophenyl)-6-fluoro-7-substituted amino-8-amino-1,4-dihydro-4-oxoquinoline-3-carboxylates 13a (mp 129-134°, 69.8%) or 13b (mp 102-104°, 57%). Cyclization of 13a and 13b in refluxing pyridine yielded the 4,12-dihydro-4-oxoquino-[1,8a,8-a,b]quinoxaline derivatives 14a (mp 233-234°, 73%) and 14b (mp 240-242°, 73%). Hydrolysis of 14a and 14b with hydrochloric acid yielded 1-(3-aminopyrrolidin-1yl)-2,9-difluoro-4,12-dihydro-4-oxoquino[1,8a,8-a,b]quinoxaline-5-carboxylic acid hydrochloride (15) (mp > 250°, 63%) and 1-(4-methylpiperazin-1-yl)-2,9-difluoro-4,12dihydro-4-oxoquino[1,8a,8-a,b]quinoxaline-5-carboxylic acid hydrochloride (16) (mp $> 250^{\circ}$, 94%).

Biological Results.

Table I summarizes the in vitro antibacterial activity of the 4,12-dihydro-4-oxoquino[1,8a,8-a,b]quinoxaline-5carboxylic acids against five gram-positive bacteria (Staphylococcus aureus ATCC 6538P, Staphylococcus aureus CMX 686B, Staphylococcus epidermidis 3519, Streptococcus faecium ATCC 8043, and Streptococcus pyogenes 930) and six gram-negative organisms (Escherichia coli Juhl, Enterobacter aerogenes ATCC 13048, Klebsiella pneumoniae 8045, Pseudomonas aeruginosa 5007, Pseudomonas aeruginosa K799/WT, and Acinetobacter sp. (MX669). The data for 1-(4-methylpiperazin-1yl)-2-fluoro-4-oxo-4H-quino[8,8a,1-a,b][1,4]benzoxazine-5-carboxylic acid (18) is included for comparison. Inspection of Table I revealed that the two compounds, 15 and 16, have moderate antibacterial activities against grampositive bacteria. However, they are less potent than their oxobiostere analogs.

Table I

In Vitro Antibacterial Activity of 4,12-dihydro-4-oxoquino[1,8a,8-a,b]quinoxaline-5-carboxylic Acids

Minimal Inhibitory Concentration (MIC), µg/ml [a]

	Organism											
Compound	Sa(A)	Sa	Se	Sf	Sp	Ec	Ea	Kp	Pa(5)	Pa(k)	A	
15	6.2	12.5	6.2	12.5	6.2	100	>100	25	>100	>100	>100	
16	6.2	6.2	3.1	3.1	0.78	12.5	50	6.2	>100	>100	12.5	
18	1.56	1.56	1.56	1.56	3.1	3.1	6.2	1.56	12.5	12.5	3.1	

[a] The MICs were detemined by the two-fold agar dilution on Brain-Heart Infusion agar. Organisms selected for inclusion in the table are: Sa(A), Staphylococcus aureus ATCC 6538P; Sa, Staphylococcus aureus CMX 686B; Se, Staphylococcus epidermidis 3519; Sf, Streptococcus faecium ATCC8043; Sp, Streptococcus pyogenes 930; Ec, Escherichia coli Juhl; Ea, Enterobacter aerogenes ATCC 13048; Kp, Klebsiella pneumoniae 8045; Pa(5), Pseudomonas aeruginosa 5007; Pa(k), Pseudomonas aeruginosa K799/WT; A, Acinetobacter sp. CMX 669.

EXPERIMENTAL

Melting points were taken in a Thomas-Hoover capillary apparatus and are uncorrected. Elemental analyses were obtained (unless otherwise specified) were within ±0.4% of the theoretical values. Microanalyses were performed by the Abbott analytical department. The nmr spectra were obtained on a Varian T-60 and HA-100 spectrometer using tetramethylsilane as an internal standard. The nmr peaks were designated as follows: s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet; b, broad. Mass spectra were recorded on a Kratos MS-50 mass spectrometer. The ir spectra were recorded on a Perkin-Elmer Model 710 A infrared spectrometer. The ir, nmr, and ms data of all compounds were consistent with the assigned structures. Solutions were dried over magnesium sulphate. 2,4-Dichloro-3-nitro-5-fluorobenzoic acid (8).

2,4-Dichloro-5-fluoroacetophenone (7) [8] (47.4 g, 230 mmoles) was added slowly to concentrated sulfuric acid (500 ml) with stirring. The reaction temperature rose to 30° and a yellowish solution resulted. Ninety percent nitric acid (57.67 g, 920 mmoles) was added dropwise over a period of 3.5 hours. The rate of addition was controlled so that the reaction temperature was maintained at 55°. After the addition, the mixture was heated at 75° for 3 hours. The mixture was cooled to room temperature and poured into 4 liters of crushed ice. A solid appeared and was filtered and washed with water (5 x 200 ml). The solid was taken up in ether and extracted twice with dilute sodium hydroxide solution. The sodium hydroxide solution was acidified to pH 2 with concentrated hydrochloric acid. An oily precipitate appeared and was extracted with ether and dried. Evaporation of the ether under reduced pressure yielded 2,4-dichloro-3-nitro-5-fluorobenzoic acid (8) (47.9 g, 82%), mp 188-191°; 'H nmr (deuteriochloroform): δ 7.95 (d, J_{H-F} = 8 Hz, 1, aromatic H).

Anal. Calcd. for C, H₂Cl₂FNO₄: C, 33.07; H, 0.79; F, 7.48; N, 5.51. Found: C, 33.04; H, 0.83; F, 7.38; N, 5.45.

Ethyl 2,4-dichloro-3-nitro-5-fluorobenzoylacetate (9).

One drop of dimethylformamide (DMF) was added to a solution of 2,4-dichloro-3-nitro-5-fluorobenzoic (29.46 g, 116 mmoles) in thionyl chloride (100 ml). After heating the mixture at 78° for 2 hours, the solvent was removed by evaporation under reduced pressure yielding a yellowish oil, 2,4-dichloro-3-nitro-5-fluorobenzoyl chloride. Monoethyl malonate (30.9 g, 234 mmoles) and 10 mg of biquinoline were dissolved in 500 ml dry tetrahydrofuran (THF) under nitrogen atmosphere and cooled to -50°. A solution of 2.5 M n-butyllithium in hexane was added until a pink color remained at -5° (200 ml). The reaction was then cooled to -78°. The acid chloride obtained as described above, dissolved in 200 ml of THF, was added dropwise. Upon addition, the dry ice bath was removed and the reaction was allowed to warm up to room temperature. The reaction was acidified with 300 ml of 1 N hydrochloric acid and was extracted with ether. The ether fraction was washed with saturated aqueous sodium bicarbonate solution and then water. The ether fraction was dried and evaporated to dryness yielding an oil 9 (36.44 g, 97%); 'H nmr (deuteriochloroform): δ (2 set of signals) 1.27 (t, J = 7 Hz, 3, ethyl CH_3), 1.35 (t, J = 7 Hz, 3 ethyl CH_3), 4.02 (s, 1, CH_2), 4.21 (q, J = 7 Hz, 2, ethyl CH₂), 4.31 (q, J = 7 Hz, 2, ethyl CH₂), 5.62 (s, 1, vinyl H), 7.61 (d, J_{H-} $_{F}$ = 10 Hz, 1 aromatic H), 7.62 (d, J_{H-F} = 10 Hz, 1 aromatic H).

Anal. Calcd. for C₁₁H₈Cl₂FNO₅: C, 40.77; H, 2.49; N, 4.32. Found: C, 40.95; H, 2.27; N, 4.07.

Ethyl 1-(2,4-Difluorophenyl)-6-fluoro-7-chloro-8-nitro-1,4-dihydro-4-oxoquinoline-3-carboxylate (11).

A solution of ethyl 2,4-dichloro-3-nitro-5-fluorobenzoylacetate (9) (12 g, 37 mmoles) in triethylorthoformate (9.1 ml, 59.2 mmoles) and acetic anhydride (24 ml, 167 mmoles) was heated at 125° for 3 hours with the removal of ethyl acetate formed during the reaction. The solution was evaporated under reduced pressure to a mobile oil which was then dissolved in methylene chloride (150 ml). 2,4-Difluoroaniline (4.18 ml, 40.7 mmoles) was added to the solution. After 15 minutes, the solution was evaporated to dryness yielding an oil 10 which was dissolved in dry

tetrahydrofuran (100 ml). To this solution at ice bath was slowly added a 60% sodium hydride-in-oil suspension (1.69 g, 38.8 mmoles). The mixture was heated at 45° for 2 hours under nitrogen atmosphere and was cooled and evaporated under reduced pressure to approximately 10 ml. Ice cold water (500 ml) was added and the mixture was filtered. The yellowish solid was washed with water and ether and dried yielding 12.5 g of 11 (79%), mp 189.5°; 'H nmr (deuteriochloroform): δ 1.41 (t, J = 7 Hz, 3, ethyl CH₂), 7.05 (m, 2, aromatic H), 7.42 (m, 1, aromatic H), 8.27 (s, 1, vinyl H), 8.48 (d, $J_{H-F} = 8$ Hz, 1, aromatic H).

Anal. Calcd. for $C_{10}H_{10}ClF_3N_2O_5$: C, 50.65; H, 2.34; N, 6.57. Found: C, 50.46; H, 2.34; N, 6.40.

Ethyl 1-(2,4-Difluorophenyl)-6-fluoro-7-(3-acetamidopyrrolidin-1-yl)-8-nitro-1,4-dihydro-4-oxoquinoline-3-carboxylate (12a).

To a solution of the 1,4-dihydro-4-oxoquinoline-3-carboxylate 11 (1 g, 2.3 mmoles) in THF (20 ml) at room temperature was added in 3-aceta-midopyrrolidine (986 mg, 6.9 mmoles). After stirring at room temperature under nitrogen atmosphere for 20 hours, the solution was evaporated to dryness under reduced pressure. The residue was treated with water (30 ml) and filtered and the residue was purified on silica gel column to yield 983 mg of 12a (83%), mp 117°; 'H nmr (deuteriochloroform): δ 1.40 (t, J = 7 Hz, 3, ethyl CH₃), 1.85 (m, 1, CH₂), 1.95 (s, 3, ethyl CH₃), 2.24 (m, 1, CH₂), 3.06 (m, 1, NCH₂), 3.31 (m, 2, NCH₂), 3.53 (m, 1, NCH₂), 4.39 (q, J = 7 Hz, 2, ethyl CH₂), 4.51 (m, 1 NCH), 5.62 (m, 1, NH), 7.01 (m, 2, aromatic H), 7.39 (m, 1, aromatic H), 8.18 (s, 1, vinyl H), 8.30 (d, J_{H-F} = 12 Hz, 1 aromatic H).

Anal. Calcd. for $C_{24}H_{21}F_{3}N_{4}O_{6}\cdot 2\frac{1}{2}H_{2}O$: C, 51.15; H, 4.62; N, 9.95. Found: C, 50.88; H, 4.50; N, 10.32.

Ethyl 1-(2,4-Difluorophenyl)-6-fluoro-7-(4-methylpiperazin-1-yl)-8-nitro-1,4-dihydro-4-oxoquinoline-3-carboxylate (12b).

Compound 12b was prepared from 11 by using the same experimental procedure as the preparation of 12a reacting 11 with N-methylpiperazine instead of 3-acetamidopyrrolidine, 73% yield, mp 204.5; 'H nmr (deuteriochloroform): δ 1.23 (t, J = 7 Hz, 3, ethyl CH₃), 2.17 (s, 3, NCH₃), 2.30 (m, 4, NCH₂), 3.01 (m, 4, NCH₂), 4.17 (q, J = 7 Hz, 2, ethyl CH₂), 7.20 (m, 3, aromatic H), 8.18 (s, 1, vinyl H), 8.43 (d, J_{H-F} = 12 Hz, 1, aromatic H). Anal. Calcd. for C₂₃H₂₁F₃N₄O₅: C, 56.33; H, 4.29; N, 11.43. Found: C, 56.04; H, 4.27; N, 11.25.

Ethyl 1-(2,4-Difluorophenyl)-6-fluoro-7-(3-acetamidopyrrolidin-1-yl)-8-amino-1,4-dihydro-4-oxoquinoline-3-carboxylate (13a).

A solution of 12a (750 mg, 1.44 mmoles) in ethyl acetate (50 ml) was hydrogenated under a nitrogen atmosphere in the presence of 1.4 g of Raney nickel. The mixture was then filtered and the solvent was evaporated to dryness to yield 409 mg 13a (70%), mp 129-134°; 'H nmr (deuteriochloroform): δ 1.40 (t, J = 7 Hz, 3, ethyl CH₃), 1.86 (m, 1, CH₂), 2.00 (s, 3, acetyl CH₃), 2.32 (m, 1, CH₂), 2.95 (m, 1, NCH₂), 3.19 (m, 1, NCH₂), 3.22 (m, 1, NCH₂), 3.49 (m, 1, NCH₂), 4.02 (bs, 2, NH₂), 4.39 (q, J = 7 Hz, 2, ethyl CH₂), 4.56 (m, 1, NCH), 5.73 (bs, 1, NH), 7.10 (m, 2, aromatic H), 7.55 (m, 1, aromatic H), 7.65 (d, J_{H-F} = 12 Hz, 1 aromatic H), 8.28 (s, 1, vinyl H).

Anal. Calcd. for $C_{24}H_{23}F_3N_4O_4\cdot 1\frac{1}{3}$ H_2O : C, 56.20; H, 5.00; N, 10.92. Found: C, 56.23; H, 5.37; N, 10.79.

Ethyl 1-(2,4-Difluorophenyl)-6-fluoro-7-(4-methylpiperazin-1-yl)-8-amino-1,4-dihydro-4-oxoquinoline-3-carboxylate (13b).

Compound 13b was prepared from 12b using a similar experimental procedure as the preparation of 13a, 57% yield, mp $102 \cdot 104^{\circ}$; ¹H nmr (deuteriochloroform): δ 1.43 (t, J = 7 Hz, 3, ethyl CH₃), 2.39 (m, 4, NCH₂), 2.40 (s, 3, NCH₃), 2.87 (m, 4, NCH₂), 3.33 (bs, 2, NH₂), 4.40 (q, J = 7 Hz, 2, ethyl CH₂), 7.64 (m, 2, aromatic H), 8.23 (m, 1, aromatic H), 8.37 (d, J_{H-F} = 12 Hz, 1, aromatic H), 8.73 (s, 1, vinyl H).

Anal. Calcd. for C₂₉H₂₉F₅N₄O₃·H₂O: C, 57.74; H, 5.23; N, 11.72. Found: C, 57.84; H, 5.08; N, 11.71.

Ethyl 1-(3-Acetamidopyrrolidin-1-yl)-2,9-difluoro-4,12-dihydro-4-oxoquino[1,8a,8-a,b]quinoxaline-5-carbonylate (14a).

A solution of 13a (450 mg, 0.92 mmole) in pyridine (4 ml) was heated at 110° under nitrogen atmosphere for 36 hours. The solvent was removed by evaporation under reduced pressure. The residue was dissolved in methylene chloride (50 ml) and was washed with sodium bicarbonate solution. Purification by column chromatography on silica gel gave the carboxylate 14a (312 mg, 73%), mp 233-234, ¹H nmr (DMSO-d₆): δ 1.31 (t, J = 7 Hz, 3, ethyl CH₃), 1.81 (m, 1, CH₂), 1.86 (s, 3, acetyl CH₃), 2.27 (m, 1, CH₂), 2.90 (m, 1, NCH₂), 3.12 (m, 1, NCH₂), 3.20 (m, 1, NCH₂), 3.29 (m, 1, NCH₂), 4.23 (q, J = 7 Hz, 2, ethyl CH₂), 4.44 (m, 1, NCH), 6.72 (m, 1, CONH), 7.01 (m, 1, aromatic H), 7.05 (m, 1, aromatic H), 7.62 (m, 1, aromatic H), 8.18 (d, J_{H-F} = 10 Hz, 1, aromatic H), 8.79 (s, 1, vinyl H), 8.88 (s, 1, NH).

Anal. Calcd. for $C_{24}H_{22}F_2N_4O_4\cdot 1/2$ H_2O : C, 60.37; H, 4.86; N, 11.73. Found: C, 60.16; H, 4.62; N, 11.46.

Ethyl 1-(4-Methylpiperazin-1-yl)-2,9-difluoro-4,12-dihydro-4-oxoquino-[1,8a,8-a,b]quinoxaline-5-carboxylate (14b).

Compound 14b was prepared from 13b using similar experimental procedure as the preparation of 14a, 73% yield, mp 240-242°; 'H nmr (deuteriochloroform): δ 1.28 (t, J = 7 Hz, 3, ethyl CH₃), 2.29 (s, 3, NCH₃), 2.31 (m, 4, NCH₂), 2.77 (m, 4, NCH₂), 3.25 (bs, 1, NH), 4.35 (q, J = 7 Hz, 2, ethyl CH₂), 6.61 (m, 1, aromatic H), 7.26 (m, 1, aromatic H), 7.28 (d, J_{H-F} = 12 Hz, 1, aromatic H), 8.81 (s, 1, vinyl H).

Anal. Calcd. for $C_{23}H_{22}F_2N_4O_3\cdot 1/2$ H_2O : C, 61.47; H, 5.12; N, 12.47. Found: C, 61.41; H, 5.10; N, 12.40.

1-(3-Aminopyrrolidin-1-yl)-2,9-difluoro-4,12-dihydro-4-oxoquino[1,8a,8-a,b]quinoxaline-5-carboxylic Acid Hydrochloride (15).

A suspension of **14a** (250 mg, 0.533 mmole) in 6 N hydrochloric acid (2 ml) was heated at 110° for 6 hours. The solvent was removed under reduced pressure. The residue was refluxed in ethanol for 1 hour and was filtered yielding 158 mg of **15** (63%), mp >250°, ¹H nmr (DMSO-d₆): δ 2.12 (m, 1, CH₂), 2.39 (m, 1, CH₂), 3.00 (m, 1, NCH₂), 3.28 (m, 1, NCH₂), 3.47 (m, 2, NCH₂), 3.94 (m, 1, NCH), 6.79 (m, 1, aromatic), 7.16 (d, J_{H-F} = 12 Hz, 1, aromatic H), 8.36 (bs, 5, NH₃Cl, NH₂Cl), 9.02 (s, 1, vinyl H), 8.89 (bs, 1, COOH).

Anal. Calcd. for $C_{20}H_{18}Cl_2F_2N_4O_3$: C, 50.97; H, 3.85; N, 11.89. Found: C, 50.64; H, 3.79; N, 11.60.

1-(4-Methylpiperazin-1-yl)-2,9-difluoro-4,12-dihydro-4-oxoquino[1,8a,8-a,b]quinoxaline-5-carboxylic Acid Hydrochloride (16).

Hydrolysis of 14b with 6 N hydrochloric acid under similar experimental condition as the preparation of 15 yielded compound 16, 94% yield, mp > 250°, ¹H nmr (deuteriotrifluoroacetic acid): δ 3.15 (m, 4, NCH₂), 3.75 (m, 7, NCH₂, NCH₃), 7.10 (m, 1, aromatic H), 7.76 (m, 1, aromatic H), 8.57 (m, 1, aromatic H), 9.43 (s, 1, vinyl H).

Anal. Calcd. for $C_{21}H_{18}F_2N_4O_3$ ·HCl·½ H_2O : C, 55.08; H, 4.37; N, 12.24. Found: C, 54.82; H, 4.14; N, 12.53.

In Vitro Antibacterial Activity.

The in vitro antibacterial activity of the test compound was determined by conventional agar dilution procedures. The organisms were grown overnight in brain-heart infusion (BHI) broth (Difco 0037-01-6) at 36°. Two-fold dilutions of the stock solution (2,000 μ g/ml) of test compound were made in BHI agar to obtain a test concentration ranging from 200 μ g/ml-0.005 μ g/ml. The plate was inoculated with approximately 10⁴ organisms. It was then incubated at 36° for 18 hours. The minimal inhibitory concentration (MIC) was the lowest concentration of the test compound that yielded no visible growth on the plate.

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